p. 484-490

## CHE 102 - Homework - Ch 19e Reactions Mechanisms - Substitution

Score: \_\_\_\_/30

Name:			Date:	
[6 pt]	1. Answer the following questions about Substitution Reactions and the Free Radicals mechanism for explaining them.			
	(a)	What is a free radical <b>and</b> how are they formed?		
	(b)	Write a reaction illustrating the formation of a free radio	cal.	
	(c)	$\underline{\text{Define}}$ each of the 3 steps in a Free Radical reaction.		
	(d)	What are the three types of free radicals formed? Draw	pictures and label them.	
	(e) (f)	Circle the most stable Free Radical in the pictures you de What is the "Selectivity Rule" for Substitution reactions		
	2. Wha (a)	at are the 5 steps to completing a substitution reaction?	<ul><li>3. What are the 4 types of molecules that can be Substituted?</li><li>(a)</li></ul>	
	(b)		(b)	
	(c)		(c)	
	(d)		(d)	
	(e)			

[24 pt] 4. Complete the following reactions by filling in the missing reactants or products. You should answer each part in the format the reaction is presented in. Assume monosubstitutions. If multiple products are made, circle the more stable product.

(a) 
$$+ \operatorname{Cl}_2 \longrightarrow$$

(c) 
$$+ H_2O \longrightarrow$$

(e) 
$$+ H_2 \longrightarrow$$

$$(f) \qquad + HCl \longrightarrow$$

(g) 
$$+ \operatorname{Cl}_2$$

$$(h) \hspace{1cm} + HCl \hspace{1cm} -$$