

Name: _____

Date: _____

[10 pt] 1. Name the following molecules:

(a) NaBr 1(a) _____

(b) MgF₂ 1(b) _____(c) Al₂(SO₃)₃ 1(c) _____(d) Na₂O 1(d) _____

(e) MgO 1(e) _____

(f) KOH 1(f) _____

(g) Ca(C₂H₃O₂)₂ 1(g) _____(h) ZnS₂O₃ 1(h) _____(i) Al₂(SO₄)₃ 1(i) _____(j) NH₄NO₃ 1(j) _____

[10 pt] 2. Write the formula for the following molecules:

(a) Strontium Fluoride 2(a) _____

(b) Magnesium Phosphate 2(b) _____

(c) Aluminum Hydroxide 2(c) _____

(d) Lithium Sulfate 2(d) _____

(e) Magnesium Sulfite 2(e) _____

(f) Barium Permanganate 2(f) _____

(g) Barium Phosphate 2(g) _____

(h) Zinc Phosphide 2(h) _____

(i) Ammonium Thiosulfate 2(i) _____

(j) Aluminum Carbonate 2(j) _____

CHE 101 - Homework - Ch 4b