

## Extra Practice - Formula to Name - F17 - Ver. 1

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Name: \_\_\_\_\_

Class: \_\_\_\_\_

Date: \_\_\_\_\_

Answer the following questions by filling in the name or formula:

- |   |  |
|---|--|
| 1. HBr  | 21. Hg(HS) <sub>4</sub>                              |
| 2. Ni <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>  | 22. CrF <sub>3</sub>                                 |
| 3. Sc <sub>2</sub> (CrO <sub>4</sub> ) <sub>3</sub> | 23. SbPO <sub>4</sub>                                |
| 4. ScAsO <sub>4</sub>                               | 24. Rb <sub>3</sub> AsO <sub>4</sub>                 |
| 5. AlI <sub>3</sub>                                 | 25. PBr <sub>7</sub>                                 |
| 6. Fe(SO <sub>4</sub> ) <sub>2</sub>                | 26. CrS <sub>2</sub> O <sub>3</sub>                  |
| 7. Ag <sub>3</sub> P                                | 27. KI   |
| 8. SnF <sub>3</sub>                                 | 28. Sc <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub>  |
| 9. Ba <sub>3</sub> N <sub>2</sub>                   | 29. ScN  |
| 10. Co(BrO <sub>3</sub> ) <sub>3</sub>              | 30. PO <sub>7</sub>                                  |
| 11. KIO <sub>2</sub>                                | 31. B <sub>3</sub> F <sub>3</sub>                    |
| 12. AlCl <sub>3</sub>                               | 32. K <sub>3</sub> N                                 |
| 13. Cr <sub>3</sub> N <sub>4</sub>                  | 33. Cu <sub>3</sub> (AsO <sub>4</sub> ) <sub>4</sub> |
| 14. Co(ClO <sub>4</sub> ) <sub>5</sub>              | 34. Sn <sub>3</sub> P <sub>5</sub>                   |
| 15. SrF <sub>2</sub>                                | 35. Cd <sub>3</sub> (AsO <sub>4</sub> ) <sub>2</sub> |
| 16. HI  | 36. B <sub>9</sub> I <sub>8</sub>                    |
| 17. AuO   | 37. B <sub>10</sub> Cl <sub>8</sub>                  |
| 18. Au <sub>2</sub> S <sub>2</sub> O <sub>3</sub>   | 38. CCl <sub>2</sub>                                 |
| 19. N <sub>7</sub> F <sub>7</sub>                   | 39. HC <sub>2</sub> H <sub>3</sub> O <sub>2</sub>    |
| 20. Ti(MnO <sub>4</sub> ) <sub>2</sub>              | 40. S <sub>4</sub> I <sub>7</sub>                    |

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- Question 1: Hyrdobromic Acid
- Question 2: Nickel Phosphate
- Question 3: Scandium Chromate
- Question 4: Scandium Arsenate
- Question 5: Aluminium Iodide
- Question 6: Iron (IV) Sulfate
- Question 7: Silver Phosphide
- Question 8: Tin (III) Fluoride
- Question 9: Barium Nitride
- Question 10: Cobalt (III) Bromate
- Question 11: Potassium Iodite
- Question 12: Aluminium Chloride
- Question 13: Chromium (IV) Nitride
- Question 14: Cobalt (V) Perchlorate
- Question 15: Strontium Fluoride
- Question 16: Hydroiodic Acid
- Question 17: Gold (II) Oxide
- Question 18: Gold (I) Thiosulfate
- Question 19: Heptanitrogen Heptafluoride
- Question 20: Titanium (II) Permanganate
- Question 21: Mercury (IV) Hydrogen Sulfide
- Question 22: Chromium (III) Fluoride
- Question 23: Antimony (III) Phosphate
- Question 24: Rubidium Arsenate
- Question 25: Phosphorus Heptabromide
- Question 26: Chromium (II) Thiosulfate
- Question 27: Potassium Iodide
- Question 28: Scandium Sulfate
- Question 29: Scandium Nitride
- Question 30: Phosphorus Heptaoxide
- Question 31: Triboron Trifluoride
- Question 32: Potassium Nitride
- Question 33: Copper (IV) Arsenate
- Question 34: Tin (V) Phosphide
- Question 35: Cadmium Arsenate
- Question 36: Nonaboron Octaiodide
- Question 37: Decaboron Octachloride
- Question 38: Carbon Dichloride
- Question 39: Acetic Acid
- Question 40: Tetrasulfur Heptaiodide