

CHE102 - Extra Practice - Name to Formula - F17 - Ver.

Name: _____

Class: _____

Date: _____

Answer the following questions by filling in the name or formula:

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|---------------------------------|---------------------------------|
| 1. Barium Arsenate | 21. Mercury (IV) Phosphate |
| 2. Chromium (III) Nitrite | 22. Vanadium (III) Sulfate |
| 3. Chromium (VI) Bromide | 23. Hexasilicon Tetrachloride |
| 4. Hexaphosphorus Tetrafluoride | 24. Octaphosphorus Decafluoride |
| 5. Tetraboron Diiodide | 25. Triboron Heptafluoride |
| 6. Cesium Permanganate | 26. Strontium Oxide |
| 7. Cobalt (V) Nitride | 27. Tetraboron Hexabromide |
| 8. Rubidium Nitride | 28. Scandium Iodate |
| 9. Vanadium (V) Arsenate | 29. Copper (VI) Sulfate |
| 10. Heptasulfur Decaoxide | 30. Decanitrogen Decachloride |
| 11. Sulfur Tetrachloride | 31. Scandium Selenide |
| 12. Iron (I) Sulfate | 32. Iron (IV) Oxide |
| 13. Tin (II) Chromate | 33. Antimony (VI) Sulfide |
| 14. Cadmium Sulfate | 34. Cobalt (VI) Nitride |
| 15. Tin (VI) Oxide | 35. Sodium Sulfate |
| 16. Nitric Acid | 36. Arsenic (IV) Fluoride |
| 17. Antimony (V) Bromate | 37. Heptaboron Tribromide |
| 18. Copper (VI) Arsenate | 38. Copper (IV) Iodite |
| 19. Cadmium Arsenate | 39. Lithium Permanganate |
| 20. Titanium (I) Selenide | 40. Titanium (V) Oxalate |

- Question 1: $\text{Ba}_3(\text{AsO}_4)_2$
Question 2: $\text{Cr}(\text{NO}_2)_3$
Question 3: CrBr_6
Question 4: P_6F_4
Question 5: B_4I_2
Question 6: CsMnO_4
Question 7: Co_3N_5
Question 8: Rb_3N
Question 9: $\text{V}_3(\text{AsO}_4)_5$
Question 10: S_7O_{10}
Question 11: SCl_4
Question 12: Fe_2SO_4
Question 13: SnCrO_4
Question 14: CdSO_4
Question 15: SnO_3
Question 16: HNO_3
Question 17: $\text{Sb}(\text{BrO}_3)_5$
Question 18: $\text{Cu}(\text{AsO}_4)_2$
Question 19: $\text{Cd}_3(\text{AsO}_4)_2$
Question 20: Ti_2Se
Question 21: $\text{Hg}_3(\text{PO}_4)_4$
Question 22: $\text{V}_2(\text{SO}_4)_3$
Question 23: Si_6Cl_4
Question 24: P_8F_{10}
Question 25: B_3F_7
Question 26: SrO
Question 27: B_4Br_6
Question 28: $\text{Sc}(\text{IO}_3)_3$
Question 29: $\text{Cu}(\text{SO}_4)_3$
Question 30: $\text{N}_{10}\text{Cl}_{10}$
Question 31: Sc_2Se_3
Question 32: FeO_2
Question 33: SbS_3
Question 34: CoN_2
Question 35: Na_2SO_4
Question 36: AsF_4
Question 37: B_7Br_3
Question 38: $\text{Cu}(\text{IO}_2)_4$
Question 39: LiMnO_4
Question 40: $\text{Ti}_2(\text{C}_2\text{O}_4)_5$