

CHE102 - Extra Practice - Chapter 6: Name to Formula - F14 - Ver. 4

Name: _____

Class: _____

Date: _____

Answer the following questions by filling in the name or formula:

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| 1. Sodium Iodide | 21. Diphosphorus Monobromine |
| 2. Nickel Chromate | 22. Lead (VI) Sulfide |
| 3. Antimony (I) Nitrite | 23. Scandium Dichromate |
| 4. Aluminium Bromide | 24. Antimony (V) Iodide |
| 5. Nitrogen Monooxygen | 25. Pentaboron Octaiodine |
| 6. Copper (V) Chromate | 26. Titanium (III) Chloride |
| 7. Silicon Octaoxygen | 27. Tin (V) Arsenate |
| 8. Octasilicon Pentafluorine | 28. Tin (VI) Oxalate |
| 9. Strontium Oxide | 29. Iron (VI) Nitride |
| 10. Vandium (VI) Phosphate | 30. Arsenic (V) Chromate |
| 11. Rubidium Chloride | 31. Carbon Dioxide |
| 12. Vandium (I) Selenide | 32. Sodium Nitride |
| 13. Tetranitrogen Octaoxygen | 33. Manganese (I) Phosphide |
| 14. Antimony (II) Sulfate | 34. Magnesium Nitride |
| 15. Aluminium Fluoride | 35. Copper (II) Oxide |
| 16. Sulfuric Acid | 36. Titanium (VI) Chloride |
| 17. Hyrdobromic Acid | 37. Iron (III) Phosphate |
| 18. Trinitrogen Trifluorine | 38. Tetrasilicon Hexaoxygen |
| 19. Tetrasulfur Nonfluorine | 39. Potassium Phosphate |
| 20. Potassium Selenide | 40. Silver Oxide |

- Question 1: NaI
Question 2: NiCrO₄
Question 3: SbNO₂
Question 4: AlBr₃
Question 5: NO
Question 6: Cu₂(CrO₄)₅
Question 7: SiO₈
Question 8: Si₈F₅
Question 9: SrO
Question 10: V(PO₄)₂
Question 11: RbCl
Question 12: V₂Se
Question 13: N₄O₈
Question 14: SbSO₄
Question 15: AlF₃
Question 16: H₂SO₄
Question 17: HBr
Question 18: N₃F₃
Question 19: S₄F₉
Question 20: K₂Se
Question 21: P₂Br
Question 22: PbS₃
Question 23: Sc₂(Cr₂O₇)₃
Question 24: SbI₅
Question 25: B₅I₈
Question 26: TiCl₃
Question 27: Sn₃(AsO₄)₅
Question 28: Sn(C₂O₄)₃
Question 29: FeN₂
Question 30: As₂(CrO₄)₅
Question 31: CO₂
Question 32: Na₃N
Question 33: Mn₃P
Question 34: Mg₃N₂
Question 35: CuO
Question 36: TiCl₆
Question 37: FePO₄
Question 38: Si₄O₆
Question 39: K₃PO₄
Question 40: Ag₂O