

CHE102 - Extra Practice - Chapter 6: Name to Formula - F14 - Ver. 2

Name: _____

Class: _____

Date: _____

Answer the following questions by filling in the name or formula:

- | | |
|----------------------------|------------------------------------|
| 1. Lithium Arsenate | 21. Nonanitrogen Heptaoxygen |
| 2. Potassium Sulfide | 22. Iron (II) Thiosulfate |
| 3. Cobalt (VI) Oxide | 23. Rubidium Cyanide |
| 4. Cobalt (III) Arsenate | 24. Barium Selenide |
| 5. Arsenic (II) Chloride | 25. Heptaboron Trioxxygen |
| 6. Tin (VI) Arsenate | 26. Sulfuric Acid |
| 7. Cobalt (III) Nitride | 27. Carbon Trichlorine |
| 8. Copper (II) Fluoride | 28. Sodium Arsenate |
| 9. Potassium Carbonate | 29. Antimony (IV) Hydrogen Sulfide |
| 10. Lead (V) Acetate | 30. Nonaboron Pentafluorine |
| 11. Octasilicon Difluorine | 31. Hydrochloric Acid |
| 12. Strontium Arsenate | 32. Cadmium Arsenate |
| 13. Hydroiodic Acid | 33. Dicarbon Nonabromine |
| 14. Lead (II) Permanganate | 34. Decanitrogen Decaiodine |
| 15. Tin (III) Oxide | 35. Decasilicon Heptachlorine |
| 16. Barium Bromide | 36. Scandium Phosphate |
| 17. Copper (II) Oxide | 37. Aluminium Bromate |
| 18. Magnesium Chloride | 38. Cobalt (VI) Phosphide |
| 19. Nitrogen Hexachlorine | 39. Tetranitrogen Decabromine |
| 20. Lead (VI) Oxide | 40. Gold (V) Phosphide |

- Question 1: Li_3AsO_4
Question 2: K_2S
Question 3: CoO_3
Question 4: CoAsO_4
Question 5: AsCl_2
Question 6: $\text{Sn}(\text{AsO}_4)_2$
Question 7: CoN
Question 8: CuF_2
Question 9: K_2CO_3
Question 10: $\text{Pb}(\text{C}_2\text{H}_3\text{O}_2)_5$
Question 11: Si_8F_2
Question 12: $\text{Sr}_3(\text{AsO}_4)_2$
Question 13: HI
Question 14: $\text{Pb}(\text{MnO}_4)_2$
Question 15: Sn_2O_3
Question 16: BaBr_2
Question 17: CuO
Question 18: MgCl_2
Question 19: NCl_6
Question 20: PbO_3
Question 21: N_9O_7
Question 22: FeS_2O_3
Question 23: RbCN
Question 24: BaSe
Question 25: B_7O_3
Question 26: H_2SO_4
Question 27: CCl_3
Question 28: Na_3AsO_4
Question 29: $\text{Sb}(\text{HS})_4$
Question 30: B_9F_5
Question 31: HCl
Question 32: $\text{Cd}_3(\text{AsO}_4)_2$
Question 33: C_2Br_9
Question 34: $\text{N}_{10}\text{I}_{10}$
Question 35: $\text{Si}_{10}\text{Cl}_7$
Question 36: ScPO_4
Question 37: $\text{Al}(\text{BrO}_3)_3$
Question 38: CoP_2
Question 39: N_4Br_{10}
Question 40: Au_3P_5